

Version 1.2 Revision Date 07/29/2021

1.	PRODUCT AND COMPANY I	DENTIFICATION
	Product Name	Sulfur Hexafluoride
	Chemical Formula	F <sub>6</sub> S
	Molecular Weight	146.06 g/mol
	CAS No.	2551-62-4
	EC No.	219-854-2
	Supplier Address*	ISOFLEX USA PO Box 29475 San Francisco CA 94129 United States
	Telephone	+1 415-440-4433
	Fax	+1 415-563-4433
		r Infotrac/ +1 800-535-5053
	(both supplier and manufacturer)	*May include subsidiaries or affiliate companies/divisions
	Email	iusa@isoflex.com
	Website	www.isoflex.com
	Preparation Information	ISOFLEX USA Product Safety +1 415-440-4433
	HAZARDOUS IDENTIFICATIO	DN CONTRACTOR OF CONTRACTOR
	Classification of the substance or mixture:	Gases under pressure (Liquified gas), H280 Simple Asphyxiant For the full text of the H-Statements mentioned in this section, see Section 16.
	GHS label elements	A
	Pictogram	
	Precautionary Label Statements	Warning
	Hazard Statement(s)	H280: Contains gas under pressure; may explode if heated. May displace oxygen and cause rapid suffocation.
	Precautionary Statement(s)	P410 + P403: Protect from sunlight. Store in a well-ventilated place.
	Supplemental Information	n
	Contains fluorinated gr	eenhouse gases
	Other Hazards	

Asphyxiant in high concentrations Contact with liquid may cause cold burns/frostbite

## 3. COMPOSITION / INFORMATION ON INGREDIENTS

Chemical Name:	Sulfur hexafluoride
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#### 4. FIRST AID MEASURES

5.

General Advice	Consult a physician. Show this SDS to the doctor in attendance. Move out of dangerous area. Hydrofluoric (HF) acid burns require immediate and specialized first aid and medical treatment. Symptoms may be delayed up to 24 hours depending on the concentration of HF. After decontamination with water, further damage can occur due to penetration/absorption of the fluoride ion. Treatment should be directed toward binding the fluoride ion as well as the effects of the exposure. Skin exposures can be treated with a 2.5% calcium gluconate gel repeated until burning ceases. More serious skin exposures may require subcutaneous calcium gluconate except for digital areas unless the physician is experienced in this technique, due to the potential for tissue injury from increased pressure. Absorption can readily occur through the subungual areas and should be considered when undergoing contamination. Prevention of absorption of the fluoride ion in cases of ingestion can be obtained by giving milk, chewable calcium carbonate tablets or Milk of Magnesia to conscious victims. Conditions such as hypocalcemia, hypomagnesemia and cardiac arrhythmias should be monitored for, since they can occur after exposure.
Inhalation Exposure	Move person into fresh air. If not breathing, give artificial respiration. Consult a physician.
Oral Exposure	Never give anything by mouth to an unconscious person. Rinse mouth with water. Consult a physician.
Dermal Exposure	Wash exposed area with soap and plenty of water. Consult a physician. First treatment with calcium gluconate paste.
Eye Exposure	Immediately flush eyes with plenty of water.
FIREFIGHTING MEASURES	
Suitable Extinguishing Media	Use water spray, alcohol-resistant foam, dry chemical or carbon dioxide.
Special Hazards	Sulphur oxides, hydrogen fluoride
Firefighting	
Protective Equipment	In the event of a fire, wear full protective clothing and NIOSH-approved self-contained breathing apparatus with full face piece operated in the pressure demand or other positive pressure mode.
Further Information	Use water spray to cool unopened containers.

### 6. ACCIDENTAL RELEASE MEASURES

Personal Precautions

Avoid breathing vapors, mist or gas. Ensure adequate ventilation. Evacuate personnel to safe areas.

Environmental Precautions	Do not let product enter drains.
Methods for Cleaning Up	Clean up promptly by sweeping or vacuum.
Reference to Other Sections	For disposal see Section 13.

7.	HANDLING AND STORAGE	
	Handling	For precautions see Section 2.2.
	Storage	Keep in a tightly closed container in a dry and well-ventilated place.
		Contents under pressure. Do not store in glass.
		Storage class (TRGS 510): 2A: Gases

# 8. EXPOSURE CONTROLS / PERSONAL PROTECTION

Component	CAS No.	Value	Control Parameters	Basis
Sulfur hexafluoride	2551-62-4	TWA	1,000 ppm	USA. ACGIH Threshold Limit Values (TLV)
	Remarks	Asphyx	ia	
		TWA	1,000 ppm 6,000 mg/m <sup>3</sup>	USA. NIOSH Recommended Exposure Limits
		May co	ntain highly toxic sulfur pe	entafluoride as an impurity.
		TWA	1,000 ppm 6,000 mg/m <sup>3</sup>	USA. Occupational Exposure Limits (OSHA) – Table Z-1 Limits for Air Contaminants
		The val	ue in mg/m³ is approxima	te.
		PEL	1,000 ppm 6,000 mg/m <sup>3</sup>	California permissible exposure limits for chemical contaminants (Title 8, Article 107)
Personal Prote	ective Equipr	nent		
Personal Resp	irators		use a full-face respirate AXBEK (EN 14387) re- controls. If the respirate supplied air respirator.	It shows air-purifying respirators are appropriate, or with multi-purpose combination (US) or type spirator cartridges as a backup to engineering or is the sole means of protection, use a full-face Use respirators and components tested and priate government standards such as NIOSH (US)
Skin Protection	1			g. The type of protective equipment must be he concentration and amount of the dangerous fic workplace.
			glove removal techniques skin contact with this p	oves must be inspected prior to use. Use proper ue (without touching glove's outer surface) to avoid roduct. Dispose of contaminated gloves after use plicable laws and good laboratory practices. Wash
Eye Protection				protection tested and approved under appropriate OSH (US) or EN 166 (EU).

### 9. PHYSICAL AND CHEMICAL PROPERTIES

# Appearance

Form	
Odor	

Liquefied gas No data available

### Safety Data

Odor Threshold:	No data available
pH:	No data available
Melting/Freezing Point:	-50 °C (-58 °F)
Initial Boiling Point:	-64 °C (-83 °F) at 1 hPa
Flash Point:	Not applicable
Flammability (Solid, Gas):	No data available
Flammability/Explosive Limits:	No data available
Vapor Density:	5.04 (Air = 1.0)
Vapor Pressure:	29 hPa at 21.1 °C (70.0 °F) / 22,157 hPa at 20 °C (68 °F)
Relative Density:	No data available
Solubility (H <sub>2</sub> O):	No data available
Partition Coefficient:	No data available (n-octanol/water)
Auto-ignition Temperature:	No data available
Decomposition Temperature:	No data available
Viscosity:	No data available
Explosive Properties:	No data available
Oxidizing Properties:	No data available

## 10. STABILITY AND REACTIVITY

Reactivity	No data available
Chemical Sta <mark>bili</mark> ty	Stable under recommended storage conditions
Possibility of Hazardous Reactions	No data available
Conditions to Avoid	Reacts dangerously with glass
Incompatible Materials	Strong oxidizing agents, glass
Decomposition Products	Hazardous decomposition products formed under fire conditions: Sulphur oxides, hydrogen fluoride
	Other decomposition products: No data available
	In the event of fire: See section 5

## 11. TOXICOLOGICAL INFORMATION

Acute Toxicity	No data available
Inhalation Dermal	No data available No data available
LD50 Intravenous	Rabbit – 5,790 mg/kg
Skin Corrosion/Irritation	No data available
Serious Eye Damage/Eye Irritation	No data available
Respiratory or Skin Sensitization	No data available
Germ Cell Mutagenicity	No data available

## Carcinogenicity

Carcinogenicity	
IARC	No component of this product present at levels greater than or equal to 0.1% is identified as a probable, possible or confirmed human carcinogen by IARC.
NTP	No component of this product present at levels greater than or equal to 0.1% is identified as a known or anticipated carcinogen by NTP.
OSHA	No component of this product present at levels greater than or equal to 0.1% is identified as a regulated carcinogen by OSHA.
Reproductive Toxicity	No data available
Teratogenicity	No data available
Specific Target Organ Toxicity / Single Exposure	No data available
Specific Target Organ Toxicity / Repeated Exposure	No data available
Aspiration Hazard	No data available
Additional Information	RTECS: WS4900000 Fluoride ion can reduce serum calcium levels, possibly causing fatal hypocalcemia. May be harmful. Nausea, dizziness, headache, central nervous system depression may occur.
ECOLOGICAL INFORMATION	
Toxicity	No data available
Persistence and Degradability	No data available
Bioaccumulative Potential	No data available
Mobility in <mark>So</mark> il	No data available
PBT and <mark>vP</mark> vB Assessment	PBT/vPvB assessment not available as chemical safety assessment not
	required/not conducted
Other Adverse Effects	No data available
DISPOSAL CONSIDERATIONS	
Product	Offer surplus and non-recyclable solutions to a licensed disposal company. Contact a licensed professional waste disposal service to dispose of this material

Contaminated Packaging Dispose of as unused product.

### 14. TRANSPORT INFORMATION

# DOT (US)

12.

13.

Proper Shipping Name Class UN No. Reportable Quantity (RQ) Poison Inhalation Hazard Sulfur hexafluoride 2.2 1080

dispose of this material.

#### sIMDG

Proper Shipping Name Class UN No. EMS No.

#### ΙΑΤΑ

SULPHUR HEXAFLUORIDE 2.2 1080 F-C, S-V

Proper Shipping Name Class UN No.

Sulphur hexafluoride 2.2 1080

#### 15. **REGULATORY INFORMATION** SARA 302 Components

No chemicals in this material are subject to the reporting requirements of SARA Title III, Section 302.

#### This material does not contain any chemical components with known SARA 313 Components CAS numbers that exceed the threshold (De Minimis) reporting levels established by SARA Title III, Section 313.

Sudden Release of Pressure Hazard

Sulfur hexafluoride / CAS No. 2551-62-4 / Revision Date 1993-02-16

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California Prop. 65 Components

**New Jersey Right** to Know Components

SARA 311/312 Hazards

Massachusetts Right

to Know Components Pennsylvania Right

to Know Components

This product does not contain any chemicals known to the State of California to cause cancer, birth defects, or any other reproductive harm.

#### 16. **OTHER INFORMATION**

Prepared By

Issuing Date

**Revision Date** 

Revision Note

**Revision Number** 

**ISOFLEX USA** PO Box 29475 San Francisco CA 94129 **United States** February 28, 2020 July 29, 2021 2 Required review and update

#### **ISOFLEX USA's Commonly Used Abbreviations and Acronyms\***

ACGIH	American Conference of Governmental Industrial Hygienists
ADR	European Agreement Concerning the International Carriage of Dangerous Goods by Road
AICS	Australian Inventory of Chemical Substances
ALARA	As Low As Is Reasonably Achievable
AMU	Atomic Mass Unit
ANSI	American National Standards Institute
BLS	Basic Life Support
BOD5	Biochemical Oxygen Demand

BOD2 Biochemical Oxygen Demand

CAM CAS CEN CERCLA CLP COD CPR CWA DAC DOE DOT DSL EC50 ECL	Continuous Air Monitor Chemical Abstracts Service (division of the American Chemical Society) European Committee for Standardization Comprehensive Environmental Response Compensation and Liability Act Classification, Labelling and Packaging (European Union) Chemical Oxygen Demand Controlled Products Regulations (Canada) Clean Water Act (USA) Derived Air Concentration (USA) United States Department of Energy (USA) United States Department of Transportation (USA) Domestic Substances List (Canada) Half Maximal Effective Concentration Korean Existing Chemicals List
EINECS EHS	European Inventory of Existing Commercial Chemical Substances Environmentally Hazardous Substance
ELINCS	European List of Notified Chemical Substances
EMS	Emergency Response Procedures for Ships Carrying Dangerous Goods
EPA	Environmental Protection Agency (USA)
EPCRA	Emergency Planning and Community Right-To-Know Act (EPCRA) of 1986
GHS	Globally Harmonized System
HMIS IARC	Hazardous Materials Identification System (USA) International Agency for Research on Cancer
IATA	International Air Transport Association
IBC	Intermediate Bulk Containers
ICAO	International Civil Aviation Organization
IDLH	Immediately Dangerous to Life or Health
IECSC	Inventory of Existing Chemical Substances Produced or Imported in China
IMDG	International Maritime Code for Dangerous Goods
LC50	Lethal concentration, 50 percent
LD50	Lethal dose, 50 percent
LDLO LOEC	Lethal Dose Low Lowest-Observed-Effective Concentration
MARPOL	International Convention for the Prevention of Pollution from Ships
MSHA	Mine Safety and Health Administration (USA)
NCRP	National Council on Radiation Protection & Measurements (USA)
NDSL	Non-Domestic Substances List (Canada)
NFPA	National Fire Protection Association (USA)
NIOSH	National Institute for Occupational Safety and Health (USA)
NOEC	No Observed Effect Concentration
N.O.S.	Not Otherwise Specified
	Nuclear Regulatory Commission (USA)
NTP OSHA	National Toxicology Program (USA) Occupational Safety and Health Administration (USA)
PBT	Persistent Bioaccumulative and Toxic Chemical
PEL	Permissible Exposure Limit
PICCS	Philippines Inventory of Chemicals and Chemical Substances
PIH	Poisonous by Inhalation Hazard
RCRA	Resource Conservation and Recovery Act (USA)
RCT	Radiation Control Technician
REACH	Registration, Evaluation, Authorisation and Restriction of Chemicals (Europe)
RID RQ	Regulations Concerning the International Transport of Dangerous Goods by Rail Reportable Quantity
RTECS	Registry of Toxic Effects of Chemical Substances
SARA	Superfund Amendments and Reauthorization Act (USA)
SNUR	Significant New Use Rule (TSCA)
TDG	Transportation of Dangerous Goods (Canada)
TIH	Toxic by Inhalation Hazard
TLV	Threshold Limit Value

TPQ	Threshold Planning Quantity
TSCA	Toxic Substances Control Act
TWA	Time Weighted Average
UN	United Nations (Number)
VOC	Volatile Organic Compound
vPvB	Very Persistent Very Bioaccumulative Chemical
WGK	Wassergefährdungsklassen (Germany: Water Hazard Classes)
WHMIS	Workplace Hazardous Materials Information System

\*One or more of the above-listed items may not appear in this document.

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